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EVALUATION OF NATURAL PRODUCTS AS INHIBITORS OF HUMAN IMMUNODEFICIENCY VIRUS TYPE 1 (HIV-1) REVERSE TRANSCRIPTASE¹

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ABSTRACT.-Inhibition of human'immunodeficiency virus reverse transcriptase is currently considered a useful approach in the prophylaxis and intervention of acquired immunodeficiency syndrome (AIDS), and natural products have not been extensively explored as inhibitors of this enzyme. We currently report that the reverse transcriptase assay developed for the detection of the enzyme in virions involving polyadenylic acid oligodeoxythymidylic acid (poly rA oligo dT) and radiolabeled thymidine 5'-triphosphate (TTP), can be applied as a simple method for screening the human immunodeficiency virus type 1 reverse transcriptase (HIV-1 RT) inhibitory potential of natural products. As reported herein, 156 pure natural products have been examined in this system. Benzophenanthridine alkaloids such as fagaronine chloride [1] and nitidine chloride, which are known inhibitors of avian myeloblastosis virus reverse transcriptase, demonstrated potent activity in the HIV-1 RT system, and 1 (IC₅₀ 10 μ g/ml) was adopted as a positive-control substance. Additional inhibitors found were columbamine iodide [2] and other protoberberine alkaloids, the isoquinoline alkaloid 0-methylpsychotrine sulfate [3], and the iridoid fulvoplumierin [4]. A number of indolizidine, pyrrolizidine, quinolizidine, indole, and other alkaloids, as well as compounds of many other structural classes, were tested and found to be inactive. A total of 100 plant extracts have also been evaluated, and 15 of these extracts showed significant inhibitory activity. Because tannins and other polyphenolic compounds are potent reverse transcriptase inhibitors, methods were evaluated for the removal of these from plant extracts prior to testing. Polyphenolic compounds were found to be responsible for the activity demonstrated by the majority of plant extracts. After appropriate tannin removal procedures were established, the bioassay system was shown to be generally applicable to both pure natural products and plant extracts. The method also proved useful in directing an isolation procedure with *Plumeria rubra* to yield fulvoplumierin [4] as an active compound (IC₅₀ 45 $\mu g/$ ml).

The discovery of reverse transcriptase (RT) in RNA tumor viruses demonstrated that genetic information can be transferred from RNA to DNA. The linear DNA product can then be integrated into the host genome. The RT is required for viral replication, and specific inhibitors of RT are useful antiviral agents (1).

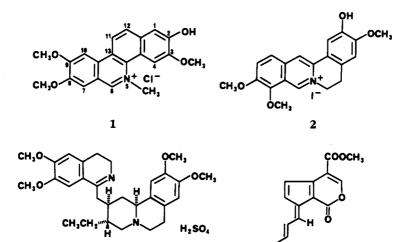
In establishing in vitro systems for screening antiviral agents, RT from RNA tumor viruses has been used by several workers to screen natural products for anti-RT activity. As a result, the benzophenanthridine and protoberberine alkaloids and certain antibiotics were shown to be potent inhibitors of the avian retrovirus RT (2–6). However, interest has shifted to HIV RT since the recognition of the causative agent of acquired immunedeficiency syndrome (AIDS) in 1981. The complex replication cycle of the human immunodeficiency virus in its host cell provides a wealth of target points for the action of drugs, including HIV RT, which is unique to the virus and is critical for viral replication. Because viral replication is essential for progression of AIDS, HIV RT repre-

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sents one of the most important targets for the development of selective antiviral compounds (7). HIV RT inhibitors such as azidothymidine (AZT), phosphonoformate (PFA), suramin, antimoniotungstate (HPA-23), and 2',3'-dideoxynucleosides have been used for pharmacological intervention of HIV, even though there are many accompanying toxicities (8). Therefore, it appears worthwhile to search for specific inhibitors of HIV RT that could demonstrate a greater therapeutic index. In addition, novel inhibitors of this enzyme may function by unique mechanisms of action.

The HIV RT assay developed for the detection of the enzyme in virions (9), using polyadenylic acid (poly A), oligodeoxythymidylic acid (oligo dT), and radiolabeled thymidine triphosphate (TTP), can be applied as a simple method for screening the HIV RT inhibitory potential of natural products. The recent availability of purified recombinant HIV-1 RT (10) makes large-scale screening of natural products for HIV RT inhibitory activity feasible. As yet, very few natural products have been shown to inhibit HIV RT specifically. Active compounds, if not eventually advocated for the treatment or prophylaxis of AIDS per se, may serve as new leads for the synthetic development of selective retroviral inhibitors.

Herein we report the development of a bioassay for the screening of natural products and plant extracts for HIV-1 RT inhibitory activity. This includes the establishment of a protocol and definitions of concentration limits for activity, solvent selection, application of the bioassay to an activity-guided fractionation procedure, and the development of strategies involved in the general screening of plant extracts. The method has also been evaluated in terms of sensitivity and selectivity.



EXPERIMENTAL

3

REAGENTS AND TEMPLATES.— $[^{3}H]$ TTP (15 Ci/mmol) was obtained from ICN Radiochemicals (Irvine, California). TTP, poly A, oligo (dT)_{12–18}, dithiothreitol, glutathione, gelatin, and soluble polyvinylpyrrolidone (PVP, average mol wt 360,000) were purchased from Sigma Chemical (St. Louis, Missouri). DEAE-cellulose filter discs (Whatman DE 81) were obtained from VWR Scientific (Batavia, Illinois). Polyamide-CC6 powder was obtained from Brinkmann Instruments (Westbury, New York). All other reagents were of analytical grade.

HIV-1 REVERSE TRANSCRIPTASE (HIV-1 RT).—The HIV-1 RT is a 66 kD recombinant enzyme obtained in an *Escherichia coli* expression system using a genetically engineered plasmid; the enzyme was purified to near homogeneity (10). Synthetic DNA segments were used to introduce initiation and termi-

nation codons into the HIV-1 RT coding sequence, and this permitted expression of large quantities of HIV-1 RT in *E. coli* (11). The enzyme was shown to be active in RT assays and exhibited inhibitory properties with several known antiretroviral agents (e.g., AZT and suramin) that are indistinguishable from the viral enzyme (12). The purified recombinant enzyme was sufficiently similar to the viral enzyme that it can be substituted for the latter in drug screening assays (10). The recombinant HIV-1 RT preparation used in all experiments had a protein concentration of 0.11 mg/ml and an activity of 238 nmol TTP incorporated per 10 min per mg protein at 37°. Prior to performing an experiment, the enzyme was diluted tenfold with buffer analogous to that used in the assay.

HIV-1 RT AssAY.—The assay mixture (final volume 100 µl) contained the following: 50 mM Tris·HCl buffer (pH 8.0), 150 mM KCl, 5 mM MgCl₂, 0.5 mM ethylene glyco-bis-(β -aminoethylether) N,N'-tetraacetic acid (EGTA), 5 mM dithiothreitol, 0.3 mM glutathione, 2.5 µg/ml bovine serum albumin, 41 µM poly A [ϵ_{260} (mM) = 7.8], 9.5 µM oligo (dT)₁₂₋₁₈ [Σ_{265} (mM) = 5.6], 0.05% Triton X-100, 20 µM TTP, and 0.5 µCi of [³H]TTP. The reaction was started by the addition of 10 µl of HIV-1 RT, and the mixture was permitted to incubate at 37° for 1 h. Reactions were terminated by the addition of 25 µl of 0.1 M EGTA followed by chilling in ice. Aliquots of each reaction mixture (100 µl) were then spotted uniformly onto circular 2.5 cm DE-81 (Whatman) filters, kept at ambient temperature for 15 min, and washed four times with 5% aqueous Na₂HPO₄·7H₂O. This was followed by two more washings with doubly distilled H₂O. Finally, the filters were thoroughly dried and subjected to scintillation counting in a nonaqueous scintillation fluid.

For testing enzyme inhibition, five serial dilutions of samples in DMSO (10 μ l) were added to the reaction mixtures prior to the addition of enzyme (10 μ l). The final DMSO concentration used was 10%. The highest concentration of pure natural products and plant extracts tested was 200 μ g/ml. Control assays were performed without the compounds or extracts, but an equivalent volume of DMSO was added. Fagaronine chloride [1] was used as the positive control substance. This compound was isolated from *Fagara xanthoxyloides* Lam. (13). Other positive control substances used were suramin (IC₅₀ 18 μ g/ml) and daunomycin (IC₅₀ 125 μ g/ml). The assay procedure and the concentration of all components were the same as that mentioned above.

SENSITIVITY OF THE HIV-1 RT ASSAY IN THE PRESENCE OF PLANT EXTRACTS.—Five concentrations of $1 (0.2-25.0 \ \mu g/ml)$ in DMSO were tested in the presence and absence of 100 $\ \mu g/ml$ of an extract derived from *Viola yedoensis* Makino. The lowest concentration of 1 was also tested in the presence of 200 $\ \mu g/ml$ of *V. yedoensis* extract. Additionally, various amounts of 1 and a constant amount of plant extract were added to each reaction mixture to simulate a final composition of 0.2%, 0.1%, 0.01%, and 0.001% (w/w) active constituent in 200 $\ \mu g/ml$ plant extract. This plant species was chosen because in preliminary studies it was devoid of HIV-1 RT inhibitory activity.

BIOACTIVITY-GUIDED FRACTIONATION PROCEDURE.—The milled bark of *Plumeria rubra* L. (Apocynaceae, 10 g) was extracted with petroleum ether (100 ml), after which it was evaporated to dryness below 40° (extract A, 0.17 g). The marc was further extracted with MeOH (100 ml), and the dried MeOH extract (extract B, 1.8 g) was then partitioned between $CHCl_3-H_2O(1:1)$. The $CHCl_3$ (extract C) and H_2O (extract D) portions were evaporated to dryness in vacuo to yield residues of 1.3 and 0.16 g, respectively. The dried extracts A, B, C, and D were assayed at a concentration of 200 µg/ml, of which only extract A showed detectable inhibitory activity (35%). Extract A was chromatographed over Si gel (5 g) and eluted with $CHCl_3$ -petroleum ether (1:1) to afford an active fraction (520 µg) which showed 70% inhibition of HIV-1 RT activity at 200 µg/ml. Fulvoplumierin [4] in this impure fraction was identified by co-tlc in three solvent systems using an authentic sample. The identity of the pure compound obtained from this fraction was confirmed by ms to be 4 (14) which has an IC_{50} value of 45 µg/ml.

POLYAMIDE COLUMN CHROMATOGRAPHY FOR TANNIN REMOVAL.—The general procedure employed was previously established by Wall *et al.* (15). Each plant extract (3 mg) was dissolved in a minimum volume of H_2O and applied to a glass column (0.6×10 cm) packed with polyamide powder (400 mg) that had been soaked in H_2O overnight. Elution was performed with H_2O (2 ml) followed by 50% MeOH (2 ml) and finally absolute MeOH (5 ml). Eluate was collected, combined, and evaporated to dryness.

PRECIPITATION OF CONTAMINATING TANNINS WITH GELATIN/NaCl SOLUTION, SOLUBLE POLYVINYLPYRROLIDONE (PVP), AND CAFFEINE.—Gelatin-NaCl solution [containing 5% (w/v) NaCl and 0.5% (w/v) gelatin] is a reagent commonly used for the qualitative detection of tannins in plant extracts (16). Tannin removal methods using soluble PVP and caffeine were conducted according to the general procedure established by Loomis and Battaile (17) and Wall *et al.* (15), respectively. Portions of each dried plant extract (2 mg) found to contain high concentration levels of tannins were dissolved in boiling H_2O (500 µl). Soluble PVP (500 µl) at concentrations between 0 and 10% (w/v) was then added, and the mixtures were vortexed and centrifuged. Similarly, gelatin/NaCl solution (500 μ l) [containing 5% (w/v) NaCl and 0 to 10% (w/v) gelatin] and caffeine at concentrations between 0 and 10% (w/v) were also employed. The final tannin removal concentrations of these agents that resulted in the plant extracts were 0– 5% (w/v) (0–50 mg/ml). The aqueous supernatant (2 mg plant extract/ml) was then tested as such for HIV-1 RT inhibitory activity yielding a reaction mixture concentration of 200 μ g/ml. The complete removal of tannins was assumed when the percent inhibition of the plant extract became constant as the concentration of PVP, gelatin, or caffeine was increased, or when insignificant inhibitory activity could be detected. Assays were also performed with varying concentrations of PVP, gelatin/NaCl solution, or caffeine to test the individual effect of these reagents on the reaction system. The plant extracts involved in all tannin removal procedures were derived from *Duabanga sonneratioides* Buch.-Ham., *Bridelia retusa* Spreng., *Gardenia coronaria* Buch.-Ham., *Careya sphaerica* Roxb., *Terminalia catappa* L., and *Terminalia alata* Heyne ex Roth. All these extracts were determined in a preliminary study to be rich in tannins using the gelatin-NaCl test.

RESULTS AND DISCUSSION

Preliminary work involved in the development of an RT bioassay included the evaluation of various experimental parameters. The effect of varying enzyme concentrations on thymidine 5'-triphosphate (TTP) incorporation was investigated. As shown in Figure 1, product formation increased in a linear manner throughout the assay period. Based on these incorporation rates, a 1 h incubation time was adopted for these experiments. Figure 2 shows that the rate of incorporation was a linear function of enzyme concentration. Thus, a 10 μ l volume (0.11 μ g of pure HIV-1 RT) was established as a suitable amount of enzyme to use in each reaction mixture.

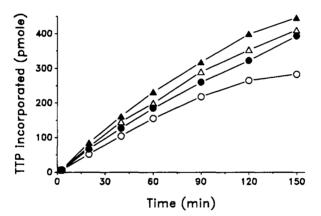


FIGURE 1. Relationship between amount of TTP incorporated and incubation time in the presence of different amounts of HIV-1 RT at 37°. Reaction mixtures were incubated with 10 (○), 12 (●), 14 (△) or 16 (▲) µl of enzyme. Aliquots (100 µl) were withdrawn at the indicated time intervals (0-150 min) and added to 25 µl of EGTA to terminate the enzyme reaction. Data points plotted are the average values obtained from duplicate experiments.

Solvents were found to exert different degrees of inhibitory activity on the polymerization reaction. As shown in Figure 3, the level of inhibition increased with increasing concentrations of DMSO, MeOH, or EtOH. These experiments demonstrated that DMSO at a 10% concentration inhibited the reaction by about 30%, whereas MeOH and EtOH at this concentration were much more inhibitory. Based on these data, DMSO (10% concentration) was selected as the solvent of choice. At this concentration, the solvent was satisfactory for the dissolution of all the test compounds used in the study.

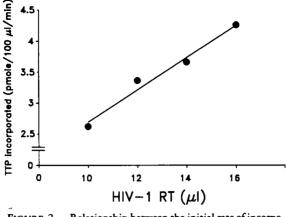
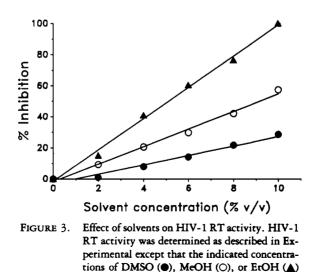


FIGURE 2. Relationship between the initial rate of incorporation (V_o) and the amount of enzyme.

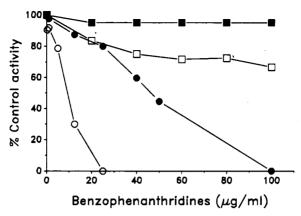
The assay procedure was found to be equally sensitive in detecting the benzophenanthridine alkaloid fagaronine chloride [1] in the presence and absence of up to 200 μ g/ml of plant extract. Thus, the percent inhibition values obtained for each concentration of 1 tested (0.2–25.0 μ g/ml) and the IC₅₀ values calculated were identical in both cases (data not shown). The lower limit of detection was found to be 0.02 μ g/ml of 1 per 200 μ g/ml of extract (i.e., 0.01% active constituent in the plant extract).

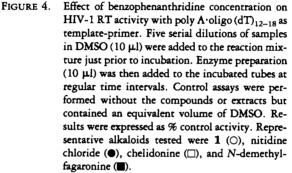
A prescieen with only one concentration of pure compound or extract (200 μ g/ml) was found to be sufficient to detect active compounds or extracts. The IC₅₀ values of all active compounds and extracts were then determined by testing five concentrations in duplicate.

Pure compounds representing many natural product classes were tested. A number of benzophenanthridine alkaloids such as 1 and nitidine chloride have shown significant cytotoxic activity against P-388 leukemia in mice (13, 18). These compounds have also been reported to be effective inhibitors of the mouse DNA polymerase and of the RTs of avian myeloblastosis virus (AMV), Rauscher murine leukemia virus, and simian sarcoma virus (19). As shown in Figure 4, 1 (IC₅₀ 10 µg/ml) and nitidine chloride (IC₅₀



were added to the incubation mixtures.





49 μ g/ml) were observed to be potent HIV-1 RT inhibitors. Chelidonine had an IC₅₀ of about 200 μ g/ml and was considered weakly active, while N-demethylfagaronine was inactive (<10% inhibition at 200 μ g/ml). Observations on the structure-activity relationships of benzophenanthridine alkaloids were analogous to those published for AMV RT (5). A hydroxyl group at position 2 enhanced activity within the class, and the optimum positions for the attachment of methoxy groups were C-8 and C-9, rather than C-7 and C-8. The quaternary nitrogen is believed to be important for anchoring such alkaloids to the double-stranded nucleic acid template (20). Although this functionality is required for the inhibitory activity of benzophenanthridine alkaloids, it is not solely responsible because quaternary alkaloids of other structural classes, e.g., echitamine chloride (an indole alkaloid) and magnoflorine chloride (an aporphine), were virtually inactive (<10% inhibition at 200 μ g/ml).

The protoberberine alkaloids are structurally related to the benzophenanthridines and have been shown to inhibit AMV RT (4). They have also been shown to possess a wide variety of biological properties such as antimicrobial and anticancer activities (21). Columbamine iodide [2] (IC₅₀ 58 μ g/ml), coptisine picrate (IC₅₀ 56 μ g/ml), jatrorrhizine chloride (IC₅₀ 71 μ g/ml), and berberine chloride (IC₅₀ 100 μ g/ml) were found to be moderately potent, whereas compounds that lacked the quaternary nitrogen, e.g., tetrahydropalmatine and tetrahydroberberine, were inactive (25% inhibition at 200 μ g/ml). The unavailability of more diversified analogues of the protoberberines prevented significant structure-activity correlation in this series of alkaloids.

The benzophenanthridine and protoberberine alkaloids are derived biosynthetically from benzylisoquinoline intermediates. The majority of alkaloids with this biosynthetic origin, however, retain the isoquinoline ring structure, as exemplified by the alkaloids from ipecac, the dried rhizome and roots of *Cephaelis ipecacuanha* (Brotero) A. Richard (Rubiaceae). Ipecac yields more than 2% w/w of Et₂O-soluble alkaloids, with the three principal ones being emetine, cephaeline, and psychotrine (22) and with 0methyl-psychotrine constituting one of the minor alkaloids (23). These structurally related alkaloids are prepared as their inorganic acid salts for solubility and stability reasons. Their pharmacological effects include antiamebic, antitumor, and antiviral activities, in addition to the irreversible inhibition of DNA synthesis (23). They also inhibit protein synthesis by inhibiting peptide elongation at the stage of translocation (23). While emetine hydrochloride, cephaeline dihydrochloride, and psychotrine free base were not active in inhibiting the HIV-1 RT (ca. 20% inhibition at 200 μ g/ml), 0methyl-psychotrine sulfate [**3**] had an IC₅₀ value of 10 μ g/ml (data not shown).

Flavonoids like tiliroside, quercetin, and kaempferol were found to be weak HIV-1 RT inhibitors (IC₅₀ 150–200 μ g/ml). This could be due to their weak intercalating properties, which are enhanced by structural planarity (24). Gossypol acetic acid, an anti-HIV compound (25), also showed weak inhibitory activity (IC₅₀ 170 μ g/ml). It is possible that this activity was due to the presence of its several phenolic hydroxy groups, because these substituents have been linked with HIV RT inhibitory activity (6).

Representatives of the quinolizidine (e.g., ormosinine and sparteine), pyrrolizidine (e.g., retrorsine) and indolizidine (e.g., swainsonine) alkaloidal groups were tested and found to be inactive (IC₅₀>200 μ g/ml). The HIV-1 RT inhibitory potential of acronycine and its derivatives (acridone alkaloids) could not be accurately assessed due to their low solubility in H₂O, resulting in precipitation in the aqueous reaction mixture. Some of the other inactive natural products tested belonged to the following classes: limonoids, lignans, sterols, quinones, triterpene glycosides, sesquiterpenes, saponin glycosides, steroidal cardiac glycosides, coumarins, rotenoids, and alkaloids belonging to the quinoline, benzylisoquinoline, aporphine, and indole types. Certain highly cytotoxic compounds were found to lack HIV-1 RT inhibitory activity, e.g., taxol (26) and didemnin B (27). Equisetin, a *Fusarium* toxin (28), was inactive despite its structural resemblance to a nucleoside.

In total, 156 compounds were tested (Table 1). A numerical summary of the results is as follows: four were active (IC₅₀<50 μ g/ml); twelve moderately active (50 μ g/ml) <IC₅₀<150 μ g/ml); five weakly active (150 μ g/ml<IC₅₀<200 μ g/ml); and 135 inactive (IC₅₀>200 μ g/ml).

TABLE 1. Compounds Tested for HIV-1 RT Inhibitory Activity.^a

Acridone alkaloids: acronycine (IA), acronycine azine (IA), dihydroacronycine (IA), nitroacronycine (IA), thioacronycine (IA)

Alkaloidal amine: colchicine (IA), ephedrine sulfate (IA), phenylephrine (IA)

Ansa macrolide: maytansine (IA)

Anthracycline antibiotic: daunomycin (MA)

Aporphine alkaloids: boldine (IA), magnoflorine chloride (IA), melosmine (IA)

Benzophenanthridine alkaloids: chelerythrine chloride (WA), chelidonine (WA), N-demethylfagaronine (IA), fagaronine chloride (A), nitidine chloride (A), sanguinarine nitrate (MA)

Cardiac glycosides: ouabain (IA)

Cephalotaxus alkaloids: harringtonine (IA), homoharringtonine (IA)

Chromone: khellin (IA)

Coumarins: esculin (IA), thamnosmonin (IA)

Diterpene alkaloids: cassaine sulfate (IA), taxol (IA)

Diterpenes/diterpene glycosides: steviol (IA), stevioside (IA)

Flavonoids: 2',4-dihydrochalcone (IA), 8-methoxy-5,7-dihydroflavonol (IA), genkwanin (IA), kaempferol (MA), quercetin (WA), tiliroside (WA), velutin (IA)

Furocoumarin: 5-(3'-methyl-2',3'-dihydroxybutyl)-8-methoxy psoralen (IA)

Furoquinoline: acetylrobustine (IA), heliparvifoline (IA)

Imidazole alkaloid: isopilosine (IA)

- Indole alkaloids: ajmalicine (IA), ajmaline (IA), alstonine hydrochloride (MA), aspidospermine (IA), aricine (IA), brucine (IA), camptothecine (IA), catharanthine hydrochloride (IA), conoflorine (IA), conopharyngine (IA), corynantheidine (IA), corynantheidine (IA), 1,2-dehydroaspidospermine (IA), deserpideine hydrochloride (IA), deserpidine (IA), echitamine chloride (IA), ellipticine (MA), iso-schizogamine (IA), isoraunescine (IA), leurosine sulfate (IA), lochnerine (IA), N-acetylperivine (IA), narceine (IA), 0-methylellipticine (MA), perakine (IA), pericalline (IA), pericyclivine (IA), periformyline (IA), perividine (IA), perivine (IA), rauniticine (IA), sarpagine (IA), schizozygine (IA), tetrahydroalstonine (IA), tubulosine (IA), vinblastine (IA), vincamajine (IA), vincristine (IA)
- Indolizidine alkaloid: swainsonine (IA)
- Iridoids: 13-0-t-coumaroylplumieride (IA), 15-demethylplumieride (IA), fulvoplumierin (A), plumericin (IA), plumieride (IA)
- Isoquinoline alkaloids: cephaeline dihydrochloride (IA), emetine hydrochloride (IA), hernandezine (IA), hydrastine (IA), 0-methylpsychotrine sulfate (A), papaverine (WA), psychotrine (IA), thalicarpine (IA), tubocurarine chloride (IA)
- Lignan: liriodendrin (IA)
- Limonoids: limonin (IA), nomilin (IA)
- Macrocyclic peptide: didemnin B (IA)
- Piperidine alkaloids: arecoline hydrobromide (IA), trigonelline hydrochloride (IA)
- Polyphenolics: ellagic acid (MA), tannic acid (A)
- Protoberberine alkaloids: berberine chloride (MA), canadine methchloride (IA), columbamine iodide (MA), coptisine chloride (MA), dihydroberberine (IA), jatrorrhizine chloride (MA), tetrahydroberberine (IA), tetrahydropalmatine (IA), tetrahydropalmatrubine (IA)
- γ-Pyrone: O-methylkojic acid (IA)
- Pyrrolizidine alkaloids: indicine N-oxide (IA), lasiocarpine (IA), monocrotaline (IA), retrorsine (IA), senecionine (IA), seneciphylline (IA)
- Quinoline alkaloids: cinchonine (IA), quinidine sulfate (IA)
- Quassinoid: bruceantin (IA)
- Quinolizidine alkaloids: nesodine (IA), ormosinine (IA), panamine (IA), panamine diperchlorate (IA), sparteine (IA), sparteine sulfate (IA), leontine (IA)
- Quinones/quinols: 2,5-dimethyl benzoquinone (IA), 3-tert-butyl-4-methoxy-1,6-dione (IA), jacaranone (IA) Rotenoids: 12-hydroxypachyrrhizone (IA), 12a-hydroxyrotenone (IA), rotenone (IA)
- Sesquiterpenes/sesquiterpene lactones: gossypol acetic acid (MA), picrotoxin (IA)
- Sesquiterpene alkaloid: phyllanthoside (IA)
- Steroids/steroidal glycosides: δ -4-androsten-3,17-dione (IA), 4 β -hydroxywithanolide (IA), polypodin (IA) Steroidal alkaloids: tomatine (IA), veratrine (IA)
- Sterol: stigmasterol (IA)
- Sulfated polysaccharide: sodium heparin (IA)
- Triterpenes: β-amyrin (IA), aridanin (IA), betulin (IA), 22-hydroxytingenone (IA), betulin (IA), primulagenin (IA)
- Triterpene saponin glycosides: ginsenoside Rb₁ (IA), ginsenoside Re (IA)

Tropane alkaloids: hyoscine hydrobromide (IA), scopolamine hydrobromide (IA)

 $^{a}A = Active$, MA = Moderately Active, WA = Weakly Active, IA = Inactive.

Approximately 100 crude EtOH or MeOH extracts from a diverse group of plant families were also tested (Table 2). Included were several which have been shown by in vitro tests to demonstrate anti-HIV activity (29), namely, Arctium lappa L., Viola yedoensis Makino, Lithospermum erythrorhizon Sieb. & Zucc., Coptis chinensis Franch., Lonicera japonica Thunb., Epimedium grandiflorum Morr., and Prunella vulgaris L. Of these, C. chinensis (IC₅₀ 108 µg/ml) was found to be significantly active. Of the 100 extracts evaluated, 15 showed >50% inhibition at 200 µg/ml and were therefore consid-

TABLE 2. Plant Extracts Tested for HIV-1 RT Inhibitory Activity.*

Annonaceae: Artabotrys siamensis Miq. (MeOH) (24%), Desmos chinensis Lour. (17%), Melodorum fruticosum Lour. (MeOH) (18%), Miliusa velutina Hook. f. & Th. (11%), Rauvenhoffia siamensis Scheff. (11%)

Apocynaceae: Allamanda violaceae Gard. & Field (22%), Beaumontia brevituba Oliv. (8%), Wrightia religiosa Benth. (8%), Alstonia macrophylla Wall. (9%), Wrightia tomentosa Roem. & Schult. (10%)

Averrhoaceae: Averrhoa carambola L. (12%)

Avicenniaceae: Avicennia alba Bl. (14%), Avicennia officinalis L. (5%)

- Barringtoniaceae: Careya sphaerica Roxb. (75%)
- Berberidaceae: Epimedium grandiflorum Morr. (15%) (IC50 = 580 µg/ml)
- Bignoniaceae: Tabebuia rosea DC. (24%), Millingtonia bortensis Linn. f. (21%), Markhamia stipulata Seem. (MeOH) (18%), Fernandoa adenophylla Steenis (16%), Radermachera ignea Steenis (12%), Stereospermum personatum Chatterjee (17%), Pachyptera hymenaea A. Gentry (9%), Dolichandrone spathacea Schum. (8%)
- Bombacaceae: Bombax ceiba L. (23%), Ceiba pentandra Gaertn. (18%)
- Boraginaceae: Litbospermum erytbrorbizon Sieb. & Zucc. (60%*) (IC50 = 212 µg/ml)
- Capparidaceae: Cleome viscosa L. (8%)
- Caprifoliaceae: Lonicera japonica Thunb. (0%) (IC50 = 1021 µg/ml)
- Casuarinaceae: Casuarina junghuhniana Miq. (48%)
- Celastraceae: Sipbonodon celastrineus Griff. (MeOH) (6%)
- Combretaceae: Terminalia catappa L. (97%*), Terminalia alata Heyne ex Roth (90%*)
- Compositae: Blumea balsamifera DC. (20%)
- Cucurbitaceae: Coccinia grandis Voigt (10%), Luffa cylindrica Roem. (12%)
- Cyperaceae: Fimbristylis quinquangularis Kunth (0%)
- Dipterocarpaceae: Shorea obtusa Wall. (12%)
- Ebenaceae: Diospyros mollis Griff. (MeOH) (11%), Diospyros gracilis Fletch. (MeOH) (30%), Diospyros montana Roxb. (MeOH) (16%), Diospyros rhodocalyx Kurz (9.4%)
- Elaeocarpaceae: Muntingia calabura L. (21%)
- Euphorbiaceae: Aporosa villosa Baill. (60%*), Sauropus androgynus Merr. (13%), Suregada multiflorum Baill. (7%), Croton crassifolius Giesel (23%), Cladogynos orientalis Zipp. ex Span. (14%), Bridelia retusa Spreng. (45%*), Breynia angustifolia Hook. f. (57%*), Drypetes roxburghii Wall. (45%*), Securinega leucopyrus Muell. Arg. (68%*)
- Flacourtiaceae: Homalium grandiflorum Benth. (8%), Homalium tomentosum Benth. (18%)
- Guttiferae: Calophyllum inophyllum L. (16%)
- Labiatae: Prunella vulgaris L. (20%) (IC30 = 385 µg/ml)
- Lauraceae: Cinnamomum iners Bl. (8%)
- Lecythidaceae: Couroupita guianensis Aubl. (MeOH) (24%)
- Leguminosae: Bauhinia winitii Craib (11%), Acacia catechu Willd. (10%)
- Meliaceae: Aglaia odorata Lour. (MeOH) (9%), Sandoricum koetjape Merr. (11%), Xylocarpus granatum Koen. (11%), Walsura trijuga Kurz (12%)
- Memecylaceae: Memecylon edule Roxb. (42%)
- Oleaceae: Myxopyrum smilacifolium Bl. (22%), Schrebera swietenioides Roxb. (53%*)
- Palmae: Salacca conferta Griff. (5%)
- Polygonaceae: Triplaris cummingiana Fisch. & Mey. (15%)
- Ranunculaceae: Coptis chinensis Franch. (94%*) (IC₅₀ = 108 µg/ml)
- Rhizophoraceae: Carallia brachiata Merr. (3%)
- Rubiaceae: Hymenodictyon excelsum Wall. (19%), Gardenia coronaria Buch.-Ham (51%*), Gardenia sootepensis Hutch. (28%), Gardenia erythroclada Kurz (9%), Morinda citrifolia L. (5%)
- Rutaceae: Murraya paniculata Jack (7%)
- Salicaceae: Salix tetrasperma Roxb. (17%)
- Salvadoraceae: Azima sarmentosa Benth. & Hook. (3%)
- Sapindaceae: Paranephelium longifoliolatum Lec. (45%*), Schleichera oleosa Merr. (8%)
- Simaroubaceae: Harrisonia perforata Merrill (MeOH) (68%*), Irvingia malayana Oliv. ex A. Benn. (37%), Ailanthus triphysa Alston (19%)
- Sonneratiaceae: Duabanga sonneratioides Buch.-Ham. (94%*)
- Staphyleaceae: Turpinia cochinchinensis Merr. (45%*)
- Stilaginaceae: Antidesma sootepense Craib (15%)
- Thymelaeaceae: Enkleia siamensis Nervling (20%)
- Ulmaceae: Celtis timorensis Span. (4%)
- Verbenaceae: Volkameria fragrans Vent. (27%)
- Violaceae: Viola yedoensis Makine (16%) (IC50 = 615 µg/ml)

 $^{\circ}$ % Inhibition values at 200 µg/ml are expressed relative to control assays and are shown in parentheses. All extracts tested were EtOH extracts except when otherwise indicated. Significantly active extracts are marked with an asterisk. ered significantly active. A response of this magnitude was considered indicative of the presence of high concentrations of potent inhibitory compounds in the plant extract, e.g., berberine (IC₅₀ 100 μ g/ml) in the roots of *Co. chinensis*. However, tannins, which appear to be very common in the plant kingdom, could also mediate a similar response. Thirteen of the active extracts gave copious precipitates with gelatin/NaCl solution, indicating that they contained high concentrations of tannins, which are known inhibitors of RTs from RNA tumor viruses (6). Also consistent with this suggestion is the fact that commercial samples of tannic and ellagic acids were found to inhibit the HIV-1 RT (80–100% inhibition at 200 μ g/ml).

Relative to tannins, most non-polyphenolic inhibitory compounds are present in much lower concentrations in plant extracts. When inhibitors such as 1, 2, and jatrorrhizine chloride were tested at concentrations between 0.002 and 0.4 μ g/ml (simulating a 0.001% to 0.2% composition in a 200 μ g/ml plant extract), only 15 to 20% inhibition was observed. Thus, in the absence of tannins, responses of this magnitude must be considered significant. However, because it is generally not known from the outset whether the inhibition mediated by a plant extract is due to tannins or other substances, chromatographic fractionation of active extracts is required, and inhibition obtained with concentrated fractions will serve as a better indication of the presence of non-tannin inhibitory compounds.

Experiments were also conducted to find quick and simple procedures for tannin removal that would not significantly affect the efficiency of the screening process. As reported previously (15), chromatography on polyamide is effective for the selective removal of polyphenolics. With slight modification, this procedure can be accomplished quickly and conveniently with only a few milligrams of plant extract. It was also observed that non-tannin inhibitory compounds with two or three phenolic hydroxy groups can be eluted by flushing with MeOH.

Precipitation of tannins with gelatin/NaCl solution was found to be an unsuitable method for tannin removal from plant extracts even though it serves as a very simple qualitative test for the presence of tannins (16). Difficulty was encountered in using the procedure quantitatively because the gelatin itself was added as an aqueous solution and there was uncertainty as to the amount to add to the plant extract for the complete precipitation of tannins. Furthermore, the viscosity of gelatin and the excess amounts in solution interfered with the enzyme assay.

Similar problems were encountered when soluble polyvinylpyrrolidone (PVP) (17) was used as a method for tannin removal. The method, however, was moderately successful because of the higher capacity of PVP for binding tannins and its less viscous nature. Percent inhibition values of plant extracts decreased as the concentration of PVP used was increased. It was found that 500 μ l of a 10% w/v PVP solution was sufficient for the complete removal of tannins from 2 mg of all the plant extracts tested, which were initially dissolved in 500 μ l of H₂O. This corresponds to an effective concentration of 50 mg/ml (5% w/v) PVP to 2 mg/ml of plant extract. When 10 μ l of PVP (50 mg/ml) was tested, there was stimulation of enzyme activity. At present, the reason for this stimulation is unknown. Nonetheless, at this stimulatory concentration, PVP did not interfere with the inhibitory effect of test compounds as demonstrated by the identical inhibition values observed in the presence and absence of PVP.

The use of caffeine (15), though free of viscosity problems, introduces a problem of solubility (ca. 20 mg/ml at 20°). A 100 mg/ml (10% w/v) aqueous solution can be readily achieved by heating; however, excess amounts recrystallized upon cooling. Because of the low capacity with which caffeine can bind to tannins, hot saturated caffeine solution (60°) must be used for plant extracts rich in polyphenolics, and 500 µl of a hot 10% w/v caffeine solution was insufficient for the complete removal of tannins from 2 mg of

all the plant extracts involved. It was also determined that the minimum inhibitory concentration of caffeine is approximately 2 mg/ml (2% w/v). Therefore, in cases where excess free caffeine exists in solution, enzyme inhibition may be observed, producing misleading results. The uncertainty as to the precise amount needed for use with each plant extract makes this method unacceptable for tannin removal in the course of sample preparation prior to screening. Also, because CHCl₃-soluble components had not been previously removed from the EtOH or MeOH plant extracts, it was not possible to remove the excess caffeine in solution by solvent extraction.

The interaction leading to the removal of tannins is the formation of hydrogen bonds between the tannin phenolic hydroxy groups and the amide link (CONH) of the precipitating agent. This interaction yields insoluble complexes. Therefore, these procedures also remove non-tannin inhibitory compounds with phenolic hydroxy groups (e.g., flavonoids), except those that have consecutive *ortho*-hydroxy groups which are internally hydrogen-bonded (although certain flavonoids could be recovered from a polyamide column by washing with MeOH). Because of this structural characteristic, hydrolyzable tannins have been reported to elude this separation process. Furthermore, quinones are irreversibly bound to these agents due to a covalent interaction (17).

Thus, no single quantitative procedure involving the addition of a precipitating reagent was found to be suitable for tannin removal from plant extracts in preparation for the HIV-1 RT assay, and no generalizations could be made regarding the concentrations or type of reagents that should be adopted. Different plant species contain different types of polyphenolics at highly variable concentrations. In our experience, the method of choice is polyamide cc, which is reliable and which does not introduce any extraneous material to the plant extract to be tested.

When the bioassay was applied at various stages of the fractionation scheme of the bark of *P. rubra*, it proved capable of directing the isolation procedure to yield fulvoplumierin [4] from the petroleum ether extract after Si gel cc. This extract showed about 35% inhibition at 200 μ g/ml, whereas the aqueous and CHCl₃ extracts were devoid of detectable inhibitory activity. Compound 4 demonstrated strong inhibitory activity with an IC₅₀ value of 45 μ g/ml (data not shown). This led to the evaluation of additional iridoids, none of which was found to be active, including plumericin and plumieride.

In conclusion, the standard reverse transcriptase assay is a specific, sensitive, simple, and reliable method for screening the HIV-1 RT inhibitory potential of natural products. Results are reproducible, and the procedure is applicable to both pure natural products and plant extracts. Moreover, the method has proven to be useful in a bioactivity-guided fractionation procedure. In general, only non-tannin inhibitory compounds are of interest. However, tannins are active inhibitors of reverse transcriptase, and no clear-cut threshold for activity could be established for plant extracts containing tannins. Tannins should be removed before the assay is performed. In the absence of tannins, a percent inhibition value of >50 at a concentration of 200 µg/ml of plant extract (an IC₅₀ \leq 200 μ g/ml) was selected as indicating potent HIV-1 RT inhibitory activity. However, a 10 to 40% inhibition value may suggest the presence of a single potent inhibitor in very low concentrations. In the case of pure natural products, >90% inhibition at 200 μ g/ml (or IC₅₀ <50 μ g/ml) was established as the threshold for potent HIV-1 RT inhibitory activity. Compounds showing percent inhibition values between 50 and 90% at 200 μ g/ml (or IC₅₀ values between 50 and 150 μ g/ml), may be classified as moderately potent. Weak inhibitors show between 30 and 50% inhibition at 200 $\mu g/ml$ (or IC₅₀ values between 150 and 200 $\mu g/ml$). Potent natural product HIV-1 RT inhibitors discovered using the present protocol include compounds from the benzophenanthridine alkaloid, protoberberine alkaloid, isoquinoline alkaloid, and iridoid classes.

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LITERATURE CITED

- 1. H.M. Temin and D. Baltimore, Adv. Virus Res., 17, 129 (1972).
- 2. V.S. Sethi, Ann. N.Y. Acad. Sci., 284, 508 (1977).
- 3. V.S. Sethi, Cancer Res., 36, 2390 (1976).
- 4. M.L. Sethi, J. Pharm. Sci., 72, 538 (1983).
- 5. M.L. Sethi, Can. J. Pharm. Sci., 16, 29 (1981).
- N. Kakiuchi, M. Hattori, T. Namba, M. Nishizawa, T. Yamagishi, and T. Okuda, J. Nat. Prod., 48, 614 (1985).
- 7. P.S. Sarin, Ann. Rev. Pharmacol., 28, 411 (1988).
- 8. E. De Clercq, J. Med. Chem., 29, 1561 (1986).
- 9. A.D. Hoffman, B. Banapour, and J.A. Levy, Virology, 147, 326 (1985).
- P.K. Clark, A.L. Ferris, D.A. Miller, A. Hizi, K.-W. Kim, S.M. Deringer-Boyer, M.L. Mellini, A.D. Clark Jr., G.F. Arnold, W.B. Lebherz III, E. Arnold, G.M. Muschik, and S.H. Hughes, *AIDS Res. Hum. Retroviruses*, 6, 753 (1990).
- 11. A. Hizi, C. McGill, and S.H. Hughes, Proc. Natl. Acad. Sci. USA, 85, 1218 (1988).
- 12. R.F. Schinazi, B.F.H. Eriksson, and S.H. Hughes, Antimicrob. Agents Chemother., 33, 115 (1989).
- 13. W.M. Messmer, M. Tin Wa, H.H.S. Fong, C. Bevelle, N.R. Farnsworth, D.J. Abraham, and J. Trojanek, J. Pharm. Sci., 61, 1858 (1972).
- L.B.S. Kardono, S. Tsauri, K. Padmawinata, J.M. Pezzuto, and A.D. Kinghorn, J. Nat. Prod., 53, 1447 (1990).
- 15. M.E. Wall, H. Taylor, L. Ambrosio, and K. Davis, J. Pharm. Sci., 58, 839 (1969).
- J.A. Wilson and H.B. Merrill, "Analysis of Leather and Materials Used in Making It," McGraw-Hill, New York, 1931, pp. 290-293.
- 17. W.D. Loomis and J. Battaile, Phytochemistry, 5, 423 (1966).
- 18. M. Tin Wa, C.L. Bell, C. Bevelle, H.H.S. Fong, and N.R. Farnsworth, J. Pharm. Sci., 63, 1476 (1974).
- 19. V.S. Sethi and M.L. Sethi, Biochem. Biophys. Res. Commun., 63, 1070 (1975).
- 20. J.M. Pezzuto, S.K. Antosiak, W.M. Messmer, M.B. Slaytor, and G.R. Honig, Chem.-Biol. Interact., 43, 323 (1983).
- 21. Y. Kondo, Heterocycles, 4, 197 (1976).
- 22. V.E. Tyler, L.R. Brady, and J.E. Robbers, "Pharmacognosy." Lea & Febiger, Philadelphia, Pennsylvania, 9th ed., 1988, pp. 209-212.
- G.A. Cordell, "Introduction to Alkaloids: A Biogenetic Approach," John Wiley & Sons, New York, 1981, pp. 560–567.
- 24. K. Shinozuka, Y. Kikuchi, C. Nishino, A. Mori, and S. Tawata, Experientia, 44, 882 (1988).
- B. Polsky, S.J. Segal, P.A. Baron, J.W.M. Gold, H. Ueno, and D. Armstrong, J. Am. Chem. Soc., 111, 8223 (1989).
- 26. M.C. Wani, H.L. Taylor, M.E. Wall, P. Coggon, and A.T. McPhail, J. Am. Chem. Soc., 93, 2325 (1971).
- 27. K.L. Rinehart, V. Kishore, K.C. Bible, R. Sakai, D.W. Sullins, and K.M. Li, J. Nat. Prod., 51, 1 (1988).
- 28. J.T. Goodwin, N.J. Phillips, and D.G. Lynn, J. Am. Chem. Soc., 111, 8223 (1989).
- 29. RS. Chang and H.W. Yeung, Antiviral Res., 9, 163 (1988).

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